

Content

IV/19 Thermodynamic Properties of Inorganic Materials

Subvolume B Binary Systems

Phase Diagrams, Phase Transition Data,
Integral and Partial Quantities of Alloys

Part 3 Binary Systems from Cs-K to Mg-Zr

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Binary Systems	1
Cs – K	3
Cs – Na	8
Cs – Rb	11
Cu – Fe	15
Cu – In	18
Cu – Li	21
Cu – Mg	24
Cu – Mn	27
Cu – Nb	32
Cu – Ni	35
Cu – O	39
Cu – Pb	41
Cu – S	44
Cu – Sb	47
Cu – Si	50
Cu – Sn	54
Cu – Sr	58
Cu – Ti	61
Cu – Tl	66
Cu – V	69
Cu – Y	72
Cu – Zn	76
Cu – Zr	81
Dy – Er	86
Dy – Ho	90
Er – Ho	94
Er – Tb	98
Fe – Mg	102
Fe – Mn	104
Fe – Mo	109
Fe – N	112
Fe – Nb	114
Fe – Nd	117

Fe – Ni	120
Fe – O	125
Fe – P	128
Fe – Pb	131
Fe – Pd	133
Fe – Pr	137
Fe – Pt	140
Fe – S	144
Fe – Si	146
Fe – Sn	149
Fe – Ti	153
Fe – V	156
Fe – W	160
Fe – Zn	163
Fe – Zr	166
Ga – Ge	170
Ga – Hg	173
Ga – In	176
Ga – P	179
Ga – Pb	181
Ga – Sb	184
Ga – Sn	187
Ga – Te	190
Ga – Zn	193
Ge – In	196
Ge – Pb	199
Ge – Sb	202
Ge – Si	205
Ge – Sn	209
Ge – Te	212
Ge – Tl	215
Ge – Zn	218
H – Nb	221
H – Zr	223
Hf – Si	226
Hf – Ta	230
Hf – Ti	234
Hg – Pb	238
Ho – Tb	241
In – P	245
In – Pb	248
In – Sb	253
In – Si	256
In – Sn	259
In – Zn	265
Ir – Pd	268
K – Rb	272
La – Ni	276
Li – Mg	280
Li – Zr	284
Mg – Mn	285
Mg – Ni	287

Mg – O	290
Mg – Sc	292
Mg – Si	297
Mg – Y	300
Mg – Zn	304
Mg – Zr	308