Contents

1.	Intr	roduction	1				
	1.1						
	1.2						
		1.2.1 First-Order Transitions Without Compositional					
		Change – The pT Phase Diagram	5				
		1.2.2 First-Order Transitions with Compositional Change					
		- The Alloy Phase Diagram	6				
		1.2.3 Second-Order Transitions	8				
	1.3 Adiabatic Decoupling of the Ionic and Electronic Degree						
		of Freedom					
1.4 The One-Electron Approximation							
	1.5 Tightly-Bound and Nearly-Free Electrons; Potentials and						
Pseudopotentials							
	1.6						
 The Statistical Mechanics of a Vibrating Lattice Periodic, Aperiodic and Quasi-Periodic Structures Elementary Excitations in Aperiodic Structures 							
						1.9.1 Effective Medium Approximations	24
						1.9.2 The Recursion Method and Related Techniques	27
	1.10 Configurational Thermodynamics of Solids and Liquids						
1.10.1 Order-Disorder Transitions			29				
		1.10.2 From the Interatomic Force Law to the Structure					
		of a Liquid	30				
		1.10.3 Order-Parameter Approach to Freezing and Melting	31				
2.	Inte	ratomic Forces in Metals and Alloys	34				
		Pseudopotentials	34				
		2.1.1 The Operator Approach	35				
		2.1.2 The Scattering Approach	37				
		2.1.3 Model Potentials	39				
	2.2 Response Theory		40				
		2.2.1 Nonlocality	45				
		2.2.2 Screening Beyond the Random Phase Approximation	46				
	2.3	Effective Pair Potentials in Pure Metals	48				

VΙΙ

	2.4	Effective Pair Potentials in Binary Alloys				
		2.4.1 Chemical Compression	58			
		2.4.2 Chemical Ordering	59			
	2.5	Interatomic Forces in Non-Simple Metals and Alloys	63			
	2.6	Beyond Perturbation Theory	65			
3.	Pha	ase Stability of Crystalline Metals	67			
	3.1	Simple-Metal Cohesion	6			
		3.1.1 An Excursion into Transition-Metal Cohesion	72			
	3.2	Structural Stability	77			
	3.3	Trends in Crystal Structures	79			
		3.3.1 A Brief Remark on Transition Metal Structures	82			
		3.3.2 The Crystal Structures of the Lanthanides	84			
		3.3.3 Charge-Density Analysis of Bonding	85			
	3.4	Pressure-Induced Phase Changes	88			
	3.5	Thermodynamics of Crystalline Metals	90			
		3.5.1 Harmonic Lattice Dynamics	90			
		3.5.2 Anharmonicity	94			
		3.5.3 Variational Method for Calculating Thermodynamic				
		Properties (Gibbs-Bogoljubov Method)	96			
	3.6	Temperature-Induced Phase Changes	98			
4.	Str	ucture and Thermodynamics of Liquid Metals	102			
	4.1		103			
	4.2	Integral Equation Approach				
	4.3	max	107			
		4.3.1 Thermodynamic Variational Method for Liquids				
		(0111 5 1111 1111 1111	107			
		400 D 11 T T T T T T T T T T T T T T T T T	110			
		4.3.3 Attractive Forces: Random Phase Approximation,				
		Optimized Random Phase Approximation and Mean				
		Spherical Approximation	113			
	4.4	Trends in Liquid Structures 121				
	4.5	Expanded Fluid Metals 125				
	4.6	Structure and Thermodynamics of Liquid Transition and				
			128			
	4.7	Atomic Motion in Liquid Metals	128			
5.	The	pT Phase Diagram of Pure Metals	1 22			
- •	5.1		133 133			
	5.2		133 137			
	5.3	TL - 1:: 13 37	137 142			
			174			

6	Alloy Formation and Stability			
	6.3	Nearly-Free-Electron Approach to Alloy Formation	145 146	
		6.1.1 Criteria for Solubility in Homovalent Systems	146	
		6.1.2 Variations of the Atomic Volume in Heterovalent	110	
		Alloy Systems	149	
		6.1.3 The Chemical Potential Model for the Heat of	110	
		Formation	154	
		6.1.4 Band Picture for Simple-Metal Alloys	157	
		6.1.5 Real-Space Picture for Alloy Formation	159	
		6.1.6 A Brief Summary	162	
	6.2	Miedema's Semiempirical Theory of Alloy Formation	163	
		6.2.1 Microscopic Interpretation of Miedema's Alloying	103	
		Rules: Simple Metals	164	
		6.2.2 Microscopic Interpretation of Miedema's Alloying	104	
		Rules: Transition Metals	165	
			103	
7.	Sol	id Substitutional Alloys	169	
	7.1	Primary Solid Solutions	171	
		7.1.1 The Homovalent Case	171	
		7.1.2 The Heterovalent Case	177	
	7.2	Hume-Rothery Phases	180	
	7.3	Static Lattice Distortions	184	
	7.4	Ordering in Substitutional Alloys		
		7.4.1 Long-Range Order	188 190	
		7.4.2 Short-Range Order	190	
		7.4.3 Ordering in Substitutional Transition-Metal Alloys.	195 196	
	7.5	Thermodynamics of Alloys	190	
		7.5.1 Lattice Dynamics of Substitutional Alloys	197	
		7.5.2 Thermodynamic Perturbation Theory	202	
		7.5.3 Vibrational Dynamics and the Ordering Transition.		
			205	
8.	Inte	ermetallic Compounds	207	
	8.1	Structure Maps	207	
;	8.2	Empirical Pair-Potential Analysis of Intermetallic Phases	211	
	8.3	Classification of Intermetallic Phases According to Building	211	
		Principles and Properties	214	
	8.4	Topologically Close-Packed Intermetallic Compounds	214	
			215	
		O 4 1 TT . 1 TT 1	215	
		0.4.0 0	215 221	
		0.4.0 0.1 70 1 1 1 01	221 227	
			230	
		8.4.5 Lattice Dynamics of Topologically Close-Packed	23U	
		<u> </u>	235	
			200	

	8.5	Intermetallic Phases with Large Band-Structure			
		Stabilization	237		
		8.5.1 Charge-Density Analysis of Bonding in Zintl Phases	238		
		8.5.2 Lattice Dynamics of Zintl Phases	239		
9.	Lia	uid Alloys	241		
	9.1				
	9.2	Thermodynamic Variational Calculations	243 246		
		9.2.1 Systems with a Nearly Ideal Mixing Behaviour	247		
		9.2.2 Liquid Alloys with Strong Chemical Short-Range	21.		
		Order	254		
		9.2.3 Liquid Alloys with a Miscibility Gap	262		
	9.3	Thermodynamic Perturbation Theory	267		
		9.3.1 Repulsive Forces: Weeks-Chandler-Andersen			
		Theory	267		
		9.3.2 Long-Range Forces: Optimized-Random-Phase			
		Approximation	272		
	9.4	Structure and Thermodynamics of Liquid Transition-Metal			
		Alloys	275		
	9.5	Collective Excitations in Liquid Alloys	278		
10	. Allo	oy Phase Diagrams	282		
	10.1	First Principles Calculations of Alloy Phase Diagrams	282		
	10.2	Chemical Short-Range Order and Alloy Phase Diagrams	285		
		10.2.1 Melting Extrema	286		
		10.2.2 Eutectic Diagrams	288		
		10.2.3 Compound Formation	288		
		10.2.4 Phase Separation in the Liquid State	289		
	10.3	Molecular Theory of the Freezing of Liquid Alloys	289		
11	Bow	ond the Phase Diagram: The Formation and			
	Pro	perties of Metastable Phases			
	111	Amorphous Alloys – Metallic Glasses	292		
		11.1.1 Glass-Forming Ability	292		
		11.1.2 Atomic Structure of Metallic Glasses	293		
		11.1.3 Elementary Excitations in Metallic Glasses	298		
	11.2	Quasi-Crystals	307		
			309		
12.	. Con	clusions and Outlook	313		
A	pend	lices	315		
Α.	Dens	sity-Functional Pseudopotentials	315		
	A.1 A.2	Optimized Pseudopotentials – the Operator Approach Norm-Conserving Pseudopotentials – the Scattering	319		
	11.6	Approach			
x		rpprometr	323		

В.	- Emour response Therby				
\mathbf{C} .	Electrostatic Energies of Crystals and Liquids				
	C.1	The N	Madelung Constants of the Elemental Structures	331	
	C.2	The N	Madelung Constants of Binary Alloys and Intermetallic		
	_	Comp	oounds	335	
_	C.3	Electr	ostatic Energies of Model Liquids and Liquid Mixtures	339	
D.	Liqu	id State	e Theory: Integral Equations, Variational Principles		
	and	Exactly	Soluble Models	341	
	D.1	Corre	lation Functions and Equations of State	341	
	D.2	Integr	al Equations and Variational Principles for the Total		
		and D	rirect Correlation Functions	343	
	D.3	Analy	tical Solutions for Model Liquids and Mixtures	350	
		D.3.1	Solution of the PY Equation for the Hard-Sphere		
		_	Fluid	350	
		D.3.2	Solution of the PY Equation for Hard-Sphere		
			Mixtures	352	
		D.3.3	The Solution of the MSA for Charged Hard Spheres		
			with Yukawa Interactions	353	
		D.3.4	The Solution of the MSA for a Symmetric Mixture		
			of Charged Hard Spheres with Yukawa Interactions	354	
Ref	oron	cos			
	CI CII	LES	•••••	357	
Sul	piect	Index	***************************************		
	J		* * * * * * * * * * * * * * * * * * * *	395	